

B.Tech/ME/3rd Sem/MECH-2101/2015

2015

APPLIED THERMODYNAMICS

(MECH 2101)

Time Alloted : 3 Hours

Full Marks : 70

Figures out of the right margin indicate full marks.

Candidates are required to answer Group A and any 5 (five) from Group B to E, taking at least one from each group.

Candidates are required to give answer in their own words as far as practicable

GROUP - A

(Multiple Choice Type Questions)

1. Choose the correct alternatives for the following : [10×1=10]

- i) For a control mass surrounded by an adiabatic wall
 - (a) entropy change > entropy generation
 - (b) entropy change < entropy generation
 - (c) entropy change = entropy generation
 - (d) none of these
- ii) In the Ideal Rankine cycle, the steam that enters the condenser leaves as a
 - (a) liquid vapour mixture
 - (b) saturated liquid
 - (c) sub cooled liquid
 - (d) superheated vapour

B.Tech/ME/3rd Sem/MECH-2101/2015

- iii) The infinitesimal and reversible work interaction in a steady flow process with no change in KE and PE is equal to
 - (a) $p dv$
 - (b) $-p dv$
 - (c) $v dp$
 - (d) $-v dp$
- iv) The value of Joule Thomson co-efficient for an ideal gas is
 - (a) positive
 - (b) negative
 - (c) zero
 - (d) can be positive or negative
- v) For the same compression ratio (r_k) & same heat rejection (Q_2)
 - (a) $\eta_{\text{Diesel}} > \eta_{\text{Dual}} > \eta_{\text{Otto}}$
 - (b) $\eta_{\text{Diesel}} < \eta_{\text{Dual}} < \eta_{\text{Otto}}$
 - (c) $\eta_{\text{Diesel}} = \eta_{\text{Dual}} = \eta_{\text{Otto}}$
 - (d) none of the above
- vi) The efficiency of a reversible cycle depends on
 - (a) nature of the working substance
 - (b) amount of the working substance
 - (c) type of the cycle followed
 - (d) temperatures of the two reservoirs between which the cycle is operated
- vii) Reheat cycles are used in vapour power plants mainly
 - (a) for increasing the efficiency
 - (b) for increasing the dryness fraction at the turbine exit
 - (c) both (a) and (b)
 - (d) none of these

- viii) In comparison with the slopes of constant pressure lines in T-s plot of an ideal gas, the slopes of constant volume lines are
- (a) more (b) less
(c) equal (d) unpredictable
- ix) _____ is not an extensive property
- (a) Volume (b) Energy
(c) Entropy (d) Pressure
- x) In a vapour compression refrigeration system, the working fluid is superheated vapour at the entrance to
- (a) evaporator (b) condenser
(c) compressor (d) expansion valve

GROUP - B

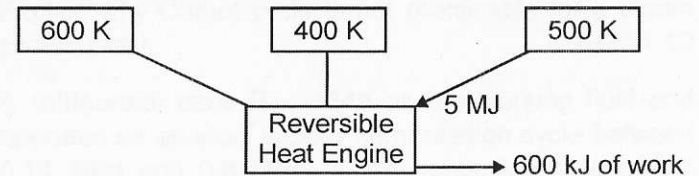
2. (a) An engine cylinder has a piston area 0.15 m^2 and contains gas at a pressure of 1.5 MPa . The gas expands according to a process which is represented by a straight line on a pressure volume diagram. The final pressure is 0.15 MPa . Calculate the work done by the gas on the piston if the stroke is 0.30 m .
- (b) A blower handles 1 kg/s of air at 20°C and consumes a power of 15 kW . The inlet and outlet velocities of air are 100 m/s and 150 m/s respectively. Find the exit air temperature assuming adiabatic conditions. Take c_p of air 1.005 kJ/kg-K . **6+6 = 12**
3. (a) A domestic refrigerator is loaded with food and the door closed. During a certain period, the machine consumes 1 kWh of energy and the internal energy of the system drops by 5000 kJ . Find the net heat transfer for the system.

- (b) A 750-L rigid tank initially contains water at 250°C , which is 50% saturated liquid and 50% saturated vapour, by volume. A small pipe connected at the top has a valve which is opened and saturated vapour is slowly withdrawn. Heat transfer takes place such that the temperature inside the tank remains constant at 250°C . Find the amount of heat transfer required to reach a stage where half the original mass is withdrawn.

4+8 = 12

GROUP - C

4. (a) A reversible heat engine, as shown in the figure draws 5 MJ heat from the 500 K reservoir and does 600 kJ of work, in a cyclic operation. Find the quantity and direction of heat interactions from the other two reservoirs.



- (b) A system maintained at constant volume is initially at temperature T_1 , and a constant temperature heat reservoir at the lower temperature T_0 is available. Show that the maximum work recoverable as the system is cooled to T_0 is $W = C_v [(T_1 - T_0) - T_0 \ln (T_1/T_0)]$; C_v is the specific heat capacity at constant volume. **6+6 = 12**
5. (a) Show that if two bodies of heat capacities C_1 and C_2 at temperatures T_1 and T_2 are brought to a temperature T by means of a reversible heat engine, then $\ln T = (C_1 \ln T_1 + C_2 \ln T_2) / (C_1 + C_2)$

- (b) A block of iron weighing 200 kg and having a temperature of 100°C is immersed in 100 kg of water at a temperature of 20°C. What will be the change of entropy of the combined system of iron and water? Specific heats of iron and water are 0.45 kJ/kg-K and 4.18 kJ/kg-K respectively. **6+6 = 12**

Group - D

6. (a) Derive the expressions for the reversible work of compression with appropriate parameters if the compression process is
- adiabatic
 - polytropic
 - isothermal
- (b) In an air standard diesel cycle, the compression ratio is 15. Compression begins at 0.1 MPa, 40°C. The heat added is 1.675 MJ/kg. Find
- the maximum temperature of the cycle
 - cut off ratio
 - cycle efficiency
 - the work done per kg of air.
- Given, $\gamma = 1.4$, $c_p = 1.005$ kJ/kg-K. **(2+1+2)+7 = 12**

7. (a) Show that the compression ratio for the maximum specific work output for an Otto cycle is given by $r_k = \left(\frac{T_{\max}}{T_{\min}} \right)^{\frac{1}{2(1-\gamma)}}$. Hence, show that the maximum specific work output is $c_v \left[\sqrt{T_{\max}} - \sqrt{T_{\min}} \right]^2$.
- (b) In an air standard Otto cycle, the compression ratio is 7, and compression begins at 33°C, 0.1 MPa. The maximum temperature of the cycle is 1127°C. Find

- the cycle efficiency the temperature and pressure at the cardinal points
 - the work done per kg of air.
- Use, $\gamma = 1.4$, $c_v = 0.718$ kJ/kg-K. **(4+3)+5 = 12**

GROUP - E

8. (a) In a reheat cycle, the initial steam pressure and the maximum temperature are 150 bar and 550°C respectively. If the condenser pressure is 0.1 bar and the moisture in the condenser inlet is 5%, determine, assuming ideal process
- the reheat pressure
 - the cycle efficiency
 - the steam rate
- (b) Explain why Carnot cycle is not practicable for a steam power plant. **8+4 = 12**
9. (a) A refrigerator uses R – 134a as the working fluid and operates on an ideal vapour compression cycle between 0.14 MPa and 0.8 MPa. If the mass flow rate of the refrigerant is 0.1 kg/s, determine
- the rate of heat removal from the refrigerated space
 - the power input
 - COP.
- Given, at 0.14 MPa: $h_g = 236.04$ kJ/kg, $s_g = 0.9322$ kJ/kg-K; at 0.8 MPa: $h_f = 93.42$ kJ/kg, enthalpy with $s = 0.9322$ kJ/kg-K is 272.05 kJ/kg;
- (b) Briefly explain the components of a vapour compression refrigeration plant. What is meant by one tonne of refrigeration? **6+4+2 = 12**