

**TRANSFER OPERATION II  
(BIOT 3104)**

**Time Allotted : 3 hrs**

**Full Marks : 70**

*Figures out of the right margin indicate full marks.*

*Candidates are required to answer Group A and any 5 (five) from Group B to E, taking at least one from each group.*

*Candidates are required to give answer in their own words as far as practicable.*

**Group – A  
(Multiple Choice Type Questions)**

1. Choose the correct alternative for the following: **10 × 1 = 10**
- (i) The process of heating a liquid mixture to form vapours and then cooling the vapours to get pure component is called \_\_\_\_\_  
(a) Crystallisation (b) Distillation  
(c) Chromatography (d) Sublimation
- (ii) For estimating the drier size it is necessary to know \_\_\_\_\_  
(a) Time of drying (b) Heat of drying  
(c) Speed of drying (d) All of the mentioned
- (iii) Which of the following is not an adsorbent?  
(a) Carbon (b) Polymers and resins  
(c) Clay (d) Dry sponge.
- (iv) Solvent extraction is more effective when the extraction is repeated with  
(a) Extra solvent (b) Large solvent  
(c) Small solvent (d) No solvent.
- (v) With increase in pressure, the relative volatility for a binary system  
(a) increases  
(b) decreases  
(c) remains same  
(d) either (a) or (b), depends on the system.
- (vi) The solubility of a gas in a liquid generally  
(a) Increases with temperature  
(b) Decreases with temperature  
(c) Decreases with pressure  
(d) Not affected by temperature and pressure.

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- (vii) NTU = the number of theoretical stages when
- (a) The operating line and the equilibrium line are straight and parallel
  - (b) The operating line and the equilibrium line intersect each other
  - (c) The minimum operating line is parallel to equilibrium line
  - (d) None of the above.
- (viii) Which of the following is not a membrane separation process?
- (a) Reverse osmosis
  - (b) Diffusion
  - (c) Ultrafiltration
  - (d) Dialysis.
- (ix) Concentration polarization occurs when
- (a) Concentration of the solute ( $C_M$ ) is more on the membrane surface than in the bulk feed solution( $C_B$ )
  - (b)  $C_M < C_B$
  - (c)  $C_M = C_B$
  - (d) None of the above.
- (x) The driving force for molecular diffusion is
- (a) Temperature gradient
  - (b) Concentration gradient
  - (c) EMF
  - (d) None of the above.

### Group – B

2. (a) Derive an expression for diffusivity in a pseudo-steady state one way diffusion process.
- (b) Prove that  $D_{AB} = D_{BA}$ .

**6 + 6 = 12**

3. A tray tower is to be designed to absorb  $SO_2$  from an air stream by using pure water at 293K. The entering gas contains 20 mol%  $SO_2$  and that leaving 2 mol% at a total pressure of 101.3 kPa. The inert air flow rate is 150 kg air/h.m<sup>2</sup>, and the entering water flow rate is 6000 kg water/h.m<sup>2</sup>. Assuming an overall tray efficiency of 25%, how many theoretical trays are needed? What should be the number of trays actually to be employed? Assume the tower to operate at 293K and equilibrium relationship is given by  $y^* = 20x$

**12**

### Group – C

4. An enriching column is fed with a feed consisting of 40 mol% benzene and 60 mol% toluene at 101.3 KPa. The feed enters at a rate 100 kmol/hr as a saturated vapour and it is desired to produce a distillate containing 90 mol% benzene. Assume that a total condenser is used and reflux ratio is 4.0 mol per mol of distillate. Determine:
- (i) Molar rate of production of distillate and residue.
  - (ii) Composition of residue
  - (iii) Number of theoretical plates.

Equilibrium Data:

x	0	0.1	0.2	0.3	0.4	0.5	0.6	0.7	0.8	0.9	0.95	1
y	0	0.205	0.369	0.502	0.618	0.706	0.789	0.849	0.907	0.955	0.978	1

**3 + 3 + 6 = 12**

5. (a) State the principle of Batch Distillation with diagram.  
 (b) Derive the Rayleigh equation for Batch Distillation.  
 (c) Define relative volatility.

**6 + 4 + 2 = 12****Group – D**

6. (a) Derive the working formula for calculation of drying time under constant rate of drying.  
 (b) Explain the primary and secondary nucleation theory for crystallisation.  
 (c) Describe the liquid- liquid extraction process . State two major application of extraction.

**4 + 4 + (2 + 2) = 12**

7. (a) State the equation and assumptions of Langmuir isotherm for adsorption.  
 (b) What is extraction factor? How is this parameter related to composition of inflow solution and composition of raffinate?

**6 + (2 + 4) = 12****Group – E**

8. Write short notes on:  
 (i) Electrodialysis  
 (ii) Reverse osmosis.

**6 + 6 = 12**

9. In a cross flow ultra filtration system used for filtration of proteins from a fermentation broth, gel resistance increases with protein concentration according to the following equation:

$$R_G = 0.5 + 0.01C, \text{ where } C \text{ is in mg/L.}$$

Pressure at the entrance system is  $P_i = 6 \text{ atm}$  and at the exit is  $P_0 = 2 \text{ atm}$ . The shell side of the filter is open to the atmosphere, resulting in  $P_f = 1 \text{ atm}$ . The membrane resistance is  $R_M = 0.5 \text{ atm}/(\text{mg}/\text{m}^2 \cdot \text{h})$ , and the protein concentration in the broth is  $C = 100 \text{ mg/L}$ . Determine:

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- (i) The pressure drop across the membrane
- (ii) Filtration flux
- (iii) Rejection coefficient of the membrane for effluent protein concentration of  $C_i = 5\text{mg/L}$ .

**4 + 4 + 4 = 12**

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