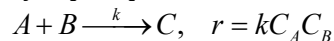


condenser area is available to maintain a pressure of 0.9492lb/in² (absolute) in the vapour space of the evaporator. On the basis of overall heat transfer coefficients of 300 Btu/hr ft² °F. Compute

- i. Heating area required and
 - ii. Steam consumption and steam economy.
- (b) Draw the temperature profile curve in a double tube heat exchanger for co-current and counter current flow.

8 + 4 = 12

7. The exothermic elementary liquid-phase reaction



is carried out in a batch reactor with a cooling coil to keep the reactor isothermal at 27°C. The reactor is initially charged with equal concentrations of A and B and no C,

Data Given: $C_{A0} = C_{B0} = 2.0 \text{ mol/L}$, $C_{C0} = 0$

- i. How long does it take to reach 95% conversion?
- ii. What is the total amount of heat (kcal) that must be removed by the cooling coil when this conversion is reached?
- iii. What is the maximum rate at which heat must be removed by the cooling coil (kcal/min) and at what time does this maximum occur?
- iv. What is the adiabatic temperature rise for this reactor and what is its significance?

12

Group - E

8. (a) Derive the expression for the modelling of electrical conductivity of a ternary mixture of carbon-polymer composite bipolar plate for PEM fuel cell.

(b) Discuss the polarization curve of a PEMFC Stack.

8 + 4 = 12

9. (a) Describe the solar thermal hot water system with internal heat exchanger built in the storage tank with schematic diagram.

(b) Explain the storage tank model without built in heat exchanger.

9 + 3 = 12

PROCESS MODELING AND SIMULATION IN ENERGY SYSTEM (REEN 5201)

Time Allotted : 3 hrs

Full Marks : 70

Figures out of the right margin indicate full marks.

Candidates are required to answer Group A and any 5 (five) from Group B to E, taking at least one from each group.

Candidates are required to give answer in their own words as far as practicable.

Group - A

(Multiple Choice Type Questions)

1. Choose the correct alternative for the following: **10 × 1 = 10**
 - (i) Distributed parameter model stands for situation
 - (a) when the process variables vary only in time coordinate
 - (b) when the process variables vary only in space coordinate
 - (c) when the process variables vary both in time and space coordinates
 - (d) none of these.
 - (ii) The overall heat transfer coefficient between steam and coolant in a condenser in case of drop wise condensation is than that of a film wise condensation.

(a) 2 to 3 times lesser	(b) 2-3 times greater
(c) 5-10 times lesser	(d) 5-10 times greater.
 - (iii) Modelling a distributed parameter system gives rise to a

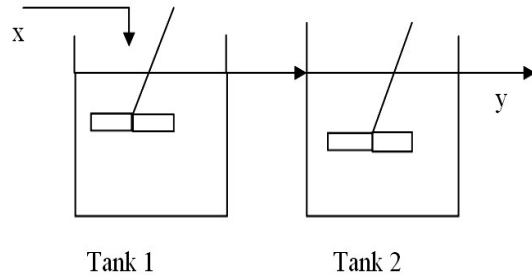
(a) ODE	(b) PDE
(c) ODE & PDE	(d) none of these.
 - (iv) The equilibrium constant of chemical reaction _____ in presence of catalyst.
 - (a) increases
 - (b) decreases
 - (c) remain unaffected
 - (d) can either increase or decrease (depends on the type of catalyst).
 - (v) Study of chemical kinetics is the easiest in case of _____ reactions.

(a) irreversible	(b) reversible	(c) surface	(d) side.
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- (vi) Lumped parameter model is represented by
 - (a) ODE
 - (b) PDE
 - (c) a set of linear equation
 - (d) none of these.
- (vii) Degree of freedom is defined as
 - (a) total number of variables
 - (b) total number of equations
 - (c) total number of variables - total number of equations
 - (d) total number of equations - total number of variables.
- (viii) For flow along a flat plate, the critical Reynold's number is
 - (a) 5×10^5
 - (b) 1×10^4
 - (c) 2×10^7
 - (d) none of these.
- (ix) Ddispersion number for a CSTR is
 - (a) 0
 - (b) 1
 - (c) < 1
 - (d) ∞ .
- (x) Which of the following type of collector is used for low temperature systems?
 - (a) Flat plate collector
 - (b) Line focussing parabolic collector
 - (c) Paraboloid dish collector
 - (d) All of the above.

Group - B

2.



In the two tank mixing process as shown in above figure, the inlet salt concentration x varies from 0 to 16 g/lit. according to a step function. The inlet flow rate is 83 lit/min. The holdup volume in each tank is 170 lit. Develop a model equation relating the outlet salt concentration of tank 2, y to the inlet salt concentration to tank 1, x assuming solution density to be constant.

At what time does the outlet salt concentration from tank 2 reaches 10 gm/lit.

12

- 3. (a) Differentiate between stochastic process and deterministic process.
- (b) The temperature of a gas stream is measured with a thermocouple. The junction may be approximated as a sphere of diameter 1 mm, thermal conductivity $k = 26 \text{ W/(m}\cdot\text{°C)}$, density $\rho = 8400 \text{ kg/m}^3$ and specific heat $c_p = 0.38 \text{ kJ/kg}$. The heat transfer coefficient between junction and the stream is $562 \text{ W/(m}^2\cdot\text{C)}$. How long will it take for the thermocouple to record 98 % of the applied temperature difference? Derive the model used.

3 + 9 = 12

Group - C

- 4. (a) A marble slab [$k = 2 \text{ W/(m}\cdot\text{°C)}$, $\alpha = 1 \times 10^{-6} \text{ m}^2/\text{s}$] 2cm thick is initially at a uniform temperature $T_i = 200\text{°C}$. Suddenly one of its surfaces is lowered to 0°C and is maintained at that temperature, while the other surface is kept insulated. Develop an explicit finite-difference scheme for the determination of the temperature distribution in the slab as a function of position and time.
- (b) State the significance of Biot no. and Fourier no.?

10 + 2 = 2

- 5. (a) Derive the explicit finite-difference form of one-dimensional, time dependent heat conduction equation.
- (b) Atmospheric air at 400K with a velocity 1.6 m/s flows over a flat plate $L = 2.2 \text{ m}$ long maintained at a uniform temperature of 300K. Calculate the heat transfer rate from the airstream to the plate if width of the plate is 0.5 m. Given: The local Nusselt no. for laminar flow along a flat plate is given by :

$$Nu_x = \frac{h_x \cdot x}{k} = 0.33 Pr^{1/3} Re_x^{1/2},$$

[For air, at 350K, $Pr = 0.7$, Kinematic viscosity = $0.2 \text{ cm}^2/\text{s}$, $k = 0.03 \text{ W/(m}\cdot\text{°C)}$.]

7 + 5 = 2

Group - D

- 6. (a) A single effect evaporator is to be designed to concentrate a 15 wt.% solution of sodium hydroxide to a 45% solution. The dilute solution at 200 °F is to be fed to the evaporator at the rate of 40,000 lb/hr. for heating purposes; a saturated steam at 50 °F is used. Sufficient