

B.TECH/ BT/6TH SEM/ BIOT 3203/2018
BIOREACTOR DESIGN AND ANALYSIS
(BIOT 3203)

Time Allotted : 3 hrs

Full Marks : 70

Figures out of the right margin indicate full marks.

*Candidates are required to answer Group A and
any 5 (five) from Group B to E, taking at least one from each group.*

*Candidates are required to give answer in their own words as far as
 practicable.*

Group - A
(Multiple Choice Type Questions)

1. Choose the correct alternative for the following: **10 × 1 = 10**
 - (i) If Thiele parameter, Φ , is greater than 3 reactor will be
 - (a) mass transfer controlling
 - (b) reaction rate controlling
 - (c) molecular diffusion controlling
 - (d) intra-particle diffusion controlling.
 - (ii) The unit of 1st order rate constant
 - (a) mol/L sec
 - (b) mol/sec
 - (c) 1/time
 - (d) 1/mol.
 - (iii) Dispersion number is given by
 - (a) D/UL
 - (b) DU/L
 - (c) DL/U
 - (d) L/DU.
 - (iv) Non-ideal reactors are characterized by
 - (a) Dispersion number
 - (b) Peclet number
 - (c) Schmidt number
 - (d) Damkohler number.
 - (v) Monod model is an equation of which of the following types?
 - (a) Linear
 - (b) Nonlinear
 - (c) Hyperbolic
 - (d) Parabolic.
 - (vi) Sherwood Number is given by the expression
 - (a) $\frac{K_L d_f}{D_{AB}}$
 - (b) $\frac{K_L d_b}{D_{AB}}$
 - (c) $\frac{K_L L_c}{D}$
 - (d) $\frac{K_L \rho}{D_{AB}}$
 - (vii) If the rate is given as $-r_A = k C_A^{0.6} C_B^{0.4}$ then the molecularity and order of the reaction is
 - (a) 1 and 1
 - (b) 1 and 2
 - (c) 2 and 1
 - (d) 2 and 2.

- (viii) Damkohler number is a measure of
 - (a) molecular diffusion
 - (b) biochemical reaction
 - (c) pore diffusion
 - (d) both (a) and (b).
- (ix) Air-lift reactor is used for production of
 - (a) alcohol
 - (b) penicillin G
 - (c) enzymes
 - (d) monoclonal antibody.
- (x) Microbial fermentation is best carried out for high yield of cell mass by
 - (a) plug flow reactor
 - (b) fed batch reactor
 - (c) back-mixed reactor
 - (d) fluidized bed reactor.

Group - B

2. (a) Derive rate equation for non-competitive inhibition (enzyme reaction) and show the result graphically.
- (b) The following set of growth data is available for a new microorganism :

t, hrs	0.0	0.5	1.0	1.5	2.0
X, dry cell mass (gm/L)	0.10	0.15	0.23	0.34	0.51

The above set of data is to be fitted to the model : $dx/dt = kx(1 - \beta x)$

Where, $\beta = 1/x_s$ and x_s is the cell mass concentration in the stationary phase. Evaluate the constants k and β .

3 + 9 = 12
3. (a) Show that the following numbers are dimensionless.
 - (i) Prandtl number
 - (ii) Schmidt number
- (b) A value of $k_{LA} = 30$ per hour has been determined for a fermentor at its maximum practical agitator rotational speed and with air being sparged at 0.5 L gas/g dry weight per hour are to be cultured. The C_{CRIT} is 0.2 mg/L. The solubility of oxygen from air in the fermenter broth is 7.3 mg/L at 30°C.
 - (i) What maximum concentration of *E.coli* can be sustained in this bioreactor under aerobic condition?
 - (ii) What concentration could be maintained if pure oxygen was used to spurge the bioreactor?
- (c) A bioreactor has an oxygen mass transfer coefficient capability of 400 h⁻¹. What is the maximum concentration of *E. coli* that can be grown aerobically in this reactor? Respiration rate of *E.coli* is 0.35 g O₂(g cell)⁻¹h⁻¹. Critical oxygen concentration is 0.2 mg/L. Assume oxygen saturation with air to be 6.7 mg/L.

2+ 6+ 4 = 12

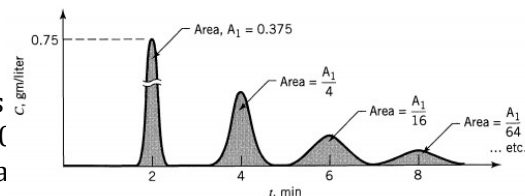
Group - C

4. (a) Find the overall order of the irreversible reaction $2\text{H}_2 + 2\text{NO} \rightarrow \text{N}_2 + 2\text{H}_2\text{O}$ from the following constant-volume data using equimolar amounts of hydrogen and nitric oxide :

Total pressure, mm Hg	200	240	280	320	360
Half-life, sec	265	186	115	104	67

- (b) Find the first-order rate constant for the disappearance of A in the gas reaction $2\text{A} \rightarrow \text{R}$ if, on holding the pressure constant, the volume of the reaction mixture, starting with 80% A, decreases by 20% in three minutes.

- 5.(a) An aqueous $\text{C}_{\text{A}0} = \text{C}_{\text{B}0} =$ (following da

**7 + 5 = 12**

ch reactor where
uation using the

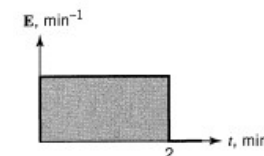
C_A (mol/L)	0.017	0.015	0.013	0.01	0.008	0.005	0.003
t , (sec)	260	490	790	1470	2205	4470	8330

- (b) Consider a gas phase, isothermal, zeroth order reaction $\text{A} \rightarrow 3\text{B}$. Initial concentration of A is 2 mol/L. Specific rate constant is 0.1 mol/L min. Final conversion required is 80%. In case of flow reactors, the volumetric flow rate to be used is 2 L/min. Calculate
- time required in a constant volume batch reactor.
 - volume required of a CSTR.
 - volume required of a PFR.

6 + 6 = 12**Group - D**

6. (a) What are the factors responsible for deviations from ideal plug flow in a tubular reactor?
- (b) Define Peclet number. What are the values of Peclet number for
- ideal CSTR.
 - ideal PFR.
- (c) A fluid reacts according to $\text{A} \rightarrow \text{R}$ as it flows through a vessel. Find the conversion of A for the flow pattern shown in the figure 1 with the following data:

$$\text{C}_{\text{A}0} = 1 \text{ mol/L}, -r_{\text{A}} = k\text{C}_{\text{A}}^{0.5}, k = 2 \text{ mol}^{0.5}/\text{L}^{0.5}.\text{min}.$$

**Figure 1****2 + 3 + 7 = 12**

7. (a) Explain the significance of (D/UL) dispersion number.
- (b) A large tank (860 liters) is used as a gas-liquid contactor. Gas bubbles up moves through the vessel and goes out through the top, liquid flows in at one part and goes out at the other at 5 L/sec. To get an idea of the flow pattern of liquid in this tank a pulse of tracer ($M = 150 \text{ gm}$) is injected at the liquid inlet and measured at the outlet, as shown below.

- Is this a properly done experiment ?
- If so, find the liquid fraction in the vessel.
- Determine the E curve for the liquid.
- Qualitatively what do you think is happening in the vessel?

2 + 10 = 12**Group - E**

8. (a) In a fluidized bed biofilm reactor with biofilms of average thickness $L=0.5 \text{ mm}$ carbon compound are to be removed. The feed flow rate and the conc. of carbon compound in the feed are $F=2 \text{ L/hr}$ and $S_0 = 2000 \text{ mg/L}$. The diameter of the column, $D=10 \text{ cm}$. The kinetic constants of the microbial population are $R_m = 50 \text{ mg/cm}^3(\text{hr})$ and $K_s = 25 \text{ mg/cm}^3$. The sp. Surface area of the biofilm in the reactor is $a = 2.5 \text{ cm}^2/\text{cm}^3$. Assume first order kinetics and effectiveness factor, $\eta = 0.7$. Determine the required height of the column, if the effluent carbon concentration $S = 100 \text{ mg/L}$.
- (b) A stirred tank reactor is to be scaled down from 10 m^3 to 0.1 m^3 . The dimension of the large tank are $D_t = 2 \text{ m}$, $D_i = 0.5 \text{ m}$, $N = 100 \text{ rpm}$. Determine the dimension of the small tank (D_t , D_i , and H) by using geometric similarity.

8 + 4 = 12

9. (a) What is a membrane bioreactor? What are the applications of membrane bioreactor? What are the major considerations of membrane bioreactor?
- (b) Discuss the importance of mass transfer coefficient and heat transfer coefficient for a bubble column reactor.

(2+2+4)+4 = 12