Odd/Sem-3	B.TECH/CHE/3 rd SEM /CHEN 2102/2015 B.TECH/CHE/3 rd SEM /CHEN 2102/2015	
the standa	Energy Engineering (CHEN 2103)	And I
OH(1) + C	Allotted : 3 hrs	F TECH
combustion OH	Figures out of the right margin indicate full marks.	101 172 040 20 0
COOH OOCCH ₃	(d) Explain the working principle of a Pitot tube. What is meant by stage pressure? (d) Explain the working principle of a Pitot tube. What is meant by stage pressure? (d) Explain the working principle of a Pitot tube. What is meant by stage pressure? (d) Explain the working principle of a Pitot tube. What is meant by stage pressure? (d) Explain the working principle of a Pitot tube. What is meant by stage pressure? (d) Explain the working principle of a Pitot tube. What is meant by stage pressure? (d) Explain the working principle of a Pitot tube. What is meant by stage pressure? (d) Explain the working principle of a Pitot tube. What is meant by stage pressure? (d) Explain the working principle of a Pitot tube. What is meant by stage pressure? (d) Explain the working principle of a Pitot tube. What is meant by stage pressure? (d) Explain the working principle of a Pitot tube. What is meant by stage pressure? (d) Explain the working principle of a Pitot tube. What is meant by stage pressure? (d) Explain the working principle of a Pitot tube. What is meant by stage pressure? (d) Explain the working principle of a Pitot tube. What is meant by stage pressure? (d) Explain tube	49 29
onoxide a	7.(a) Write down the relations between pump head, capacity and rpm for central dates are required to give answer in their own words as far as practicable. Group - A	NTARY B
° C in 90° on leave t	pump. (Multiple Choice Type Questions) 10 x 1=10	Contracts
n the react suming co	(b) A centrifugal pump has the following dimensions: Inlet diameter =160 cm; (a) Coal with higher volatile matter content has diameter = 320 cm; width of the impeller at the inlet = 5 cm; Vane angle at the (a) Higher calorific value (b) Higher ignition temperature (a) Higher tendency of spontaneous oxidation (c) None of these.	
and N_2 ar	$(\phi_1) = 0.45$ radians, and value angle at the outlet $(\phi_2) = 0.25$ radians, when $\phi_1 = 0.45$ radians, and value angle at the outlet $(B_2) = 5$ cm.	s.
.y.)	Assuming shockless entry, determine the discharge and the head developed by Coal demand in India is maximum for pump when the impeller rotates at 850 r.p.m. (a) Thermal power Industry (b) Steel Industry (d) Brick Industry.	s as far as
of liquid	5+7: (c) Cement Industry (c) Strengt gracking process?	
ugh the the the relation	Group – E (iii) Which process among the following is a thermal clacking process (b) Catalytic reforming (c) Viebreaking	
the moles	8.(a) A water softener consists of a vertical tube of 100 mm diameter and packet (c) Hydrotreating (d) Visoreaking.	
g of na	well as a pressure difference at a rate of 200 mL/s. The bed has a porosity of (a) Paraffins (b) Olefins (c) Aromatics (d) Napthenes.	
, with all	(c) A constant of protocol of finitiation is protocol of the multiplication factor must be	re
he net he	(b) Briefly describe the different types of fluidization. Explain what is measing (b) For a controlled nuclear reaction, do not be present types of fluidization. Explain what is measing (c) for a controlled nuclear reaction, do not be present types of fluidization. Explain what is measing (c) for a controlled nuclear reaction, do not be present types of fluidization. Explain what is measing (c) for a controlled nuclear reaction, do not be present types of fluidization. Explain what is measing (c) for a controlled nuclear reaction, do not be present types of fluidization. Explain what is measing (c) for a controlled nuclear reaction, do not be present types of fluidization. Explain what is measing (c) for a controlled nuclear reaction, do not be present types of fluidization. Explain what is measing (c) for a controlled nuclear reaction, do not be present types of fluidization. Explain what is measing (c) for a controlled nuclear reaction, do not be present types of fluidization. Explain what is measing (c) for a controlled nuclear reaction, do not be present types of fluidization. Explain what is measing (c) for a controlled nuclear reaction, do not be present types of fluidization. Explain what is measing (c) for a controlled nuclear reaction, do not be present types of fluidization. Explain types of fluidization is the fluidization of fluidization of fluidization. Explain types of fluidization of	
it heat of	(c)Estimate the terminal velocity for limestone particles of diameter 0.08 mm(def 2800 kg/m ³) falling in water at 30 °C. The viscosity of water is 10 ⁻³ Pascal sec. (v) Volatile matter in coal is determined by heating the coal sample in muffle furna-	ce gion of Bio nois
nitions neld of	(d) Explain the significance of void fraction during flow of fluid through packed b^{e} (a) at 800°C for 10 min (c) at 900°C for 8 min (c) at 900°C for 8 min (d) at 925°C for 7 min.	
hogan	(vii) Cetane number is highest for	
rodunos	9.(a) Explain the significance of the concept of Prandtl's mixing length theory in the concept of Prandtl's mixing length. (b) reference (c) r	
	 (b) The air is flowing over a cylinder of diameter 50 mm and infinite length w [Viii] The cold weather performance of a fuel is indicated by its velocity of 0.1 m/s. Find the total drag, shear drag and pressure drag on 1 m [e] (a) Calorific value (b) Pour point (d) Smoke point. (c) Flash point (d) Smoke point. 	
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em/CIIE/Odd/Sem-31		
	B.TECH/CHE/3rd SEM /CHEN 2103/2015	CHICHE/3" SEM /CHEN 2103/2015
Calculate the start	(ix) Which of the following has the highest calorific value?	Group – D
$C_2H_5OH(1) + C$	(c) Carburetted water gas (d) Bio gas.	Describe the operation of an anaerobic digester for producing bio-gas.
Heats of combustion C ₂ H ₅ OH	(x) A solar cell converts solar energy to (a) Chemical energy (b) Electrical energy	What is coke oven gas? How is it produced? $6+(1+5) = 12$
CH ₃ COOH C ₂ H ₅ OOCCH ₂	(c) Heat energy (d) None of these.	Write a note on Integrated gasification combined cycle.
Carbon monovide et	Group – B	compare the properties of blast furnace gas and water gas. State the uses of natural
air at 500° C in 90% combustion leave t	2.(a) What do you mean by washing of coal? What are the objectives of What is the working principle of a Jig washer? What are the advar washed coal?	c_{01} gas. 6+(3+3) = 12
evolved in the react		Group – E
burned assuming con O_2 , O_2 and N_2 are	(b) The following data are available from a coal washery using Indian coa Ash in feed coal = 21% Ash in clean product = 18.7%	l; (a) Explain the principle of operation of flat plate solar collectors. What is tracking?
espectively.)	Ash in sink = 41% If the theoretical recovery is 91.5% for a product of same ash	(b) What are heliostats? How do they work? (5+3) + (1+3) = 12
0 moles of liquid	performance of the washery.	to the sector Conversion
aks through the v	(2+2+3	3+1(a) Explain the concept of Ocean thermal Energy Conversion.
onditions the relativ	3.(a) Describe how coal is carbonized in a by-product slot type coke oven.	(b) Describe the operation of a breeder reactor. $6 + 6 = 12$
omposition is 50 mc	(b) With a neat flow diagram describe how valuable by-products are r coke oven gas.	econ
hen 1.0g of nag		· A server and a server of a server ser
ing condensed, 40.	Group – C	
lue and the net hea The latent heat of	4.(a) Why is vacuum distillation of crude oil required? State the products distillation unit in a refinery.	; frø
and Foctooffic die	(b) With respect to Fluid Catalytic Cracking process state the following: i) Feedstock used	
centere yield of	ii) Catalyst used	
Gaz Arig harstein	iii) Process conditions	
Sel nonizodmos	v) Draw the flow-sheet of the process.	
		[1 + ³
no brace on	5.(a) What is catalytic reforming? Describe the main chemical reactions that reforming. State the operating conditions of a typical reforming processing the state of the operating conditions of a typical reforming processing the state of t	it o ⁰ iss.
NORA AND BUT FOR	(b) Write a short note on visbreaking.	
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