

**ENVIRONMENTAL SCIENCES
(EVS2016)**

Time Allotted : 2½ hrs

Full Marks : 60

Figures out of the right margin indicate full marks.

*Candidates are required to answer Group A and
any 4 (four) from Group B to E, taking one from each group.*

Candidates are required to give answer in their own words as far as practicable.

Group – A

1. Answer any twelve:

12 × 1 = 12

Choose the correct alternative for the following

- (i) The Environmental Protection Act of India was enacted in the year
(a) 1974 (b) 1986
(c) 1997 (d) 1965
- (ii) Which one of the following is a green solvent?
(a) Benzene (b) Chloroform
(c) Ethyl Lactate (d) Acetone
- (iii) Which is an example of inexhaustible energy source?
(a) Coal (b) Petroleum
(c) Biomass (d) Mineral
- (iv) Air pollutant which reduces oxygen carrying capacity of haemoglobin is
(a) Carbon monoxide (b) Ammonia
(c) Carbon dioxide (d) Hydrogen sulphide
- (v) The main product of photochemical smog is
(a) PAN (b) O₃
(c) H₂SO₄ (d) NH₄Cl
- (vi) The atmosphere is stable under condition of
(a) ELR>ALR (b) ELR<ALR
(c) ELR=ALR (d) None of these
- (vii) Water will be considered as saline if the TDS value is
(a) < 500 mg/L (b) < 1500 mg/L
(c) equal to 3000 mg/L (d) > 5000 mg/L
- (viii) Lakes with poor nutrient status are known as
(a) Mesotrophic (b) Oligotrophic
(c) Eutrophic (d) None of these

- (ix) The noise threshold limit values for 16 hours is
 (a) 80 Dba (b) 90 dBA
 (c) 100 dBA (d) 110 dBA
- (x) Hazardous waste should have characteristics such as
 (a) Reactivity (b) Toxicity
 (c) Corrosivity (d) All the above

Fill in the blanks with the correct word

- (xi) The first room temperature ionic liquid is _____
- (xii) The value of earth's albedo is _____.
- (xiii) The purpose of the secondary treatment is to remove mainly _____ from the waste water.
- (xiv) In the industrial area noise is measured by _____.
- (xv) The unit of intensity of sound is _____.

Group - B

2. (a) Following exponential growth model show that doubling time and half life time of a population will be numerically equal when the growth rate constant and decay rate constant have same value. [[CO5](Apply/IOCQ)]
- (b) What are the main objectives of Environmental Protection Act of India 1986? [[CO6](Remember/LOCQ)]
- (c) What do you mean by green solvent? Differentiate between safe and green solvents giving suitable examples. [[CO6](Remember/LOCQ)]
- 4 + 3 + (2 + 3) = 12**
3. (a) What is logistic growth of population? The human population follow a logistic growth rate until it stabilizes at 10 billion. In the year 1970, the world population was 2 billion with growth rate of 2 %. When will the population reach 5 billion? [[CO5](Apply/IOCQ)]
- (b) What is renewable resource? Classify renewable resources providing suitable examples. [[CO6](Remember/LOCQ)]
- (c) What are the VOCs? Mention the harmful effects of VOCs on human health. [[CO6](Remember/LOCQ)]
- (2 + 3) + (1 + 3) + (1 + 2) = 12**

Group - C

4. (a) Why the temperature decreases with increase of height in troposphere? Mention minimum temperature achieved in troposphere. [[CO1](Apply/IOCQ)]
- (b) Why are CO₂ and water vapour (H₂O) greenhouse gases, but N₂ and O₂ are not? [[CO3](Remember/LOCQ)]
- (c) Write down the sources of generation of carbon monoxide (CO) and oxides of nitrogen (NO_x). Also describe their harmful effect on human being. [[CO2](Remember/LOCQ)]

- (d) Write short note about on 'Baghouse filter'. [[CO6](Remember/LOCQ)]
(2 + 1) + 2 + (2 + 2) + 3 = 12
5. (a) Consider the sun as a perfect sphere of radius 6.8×10^8 m. Calculate the energy radiated by the sun in 12 hours. Surface temperature of sun is 6200K and Stefan's constant $\sigma = 5.67 \times 10^{-8} \text{ Jm}^{-2}\text{s}^{-1}\text{K}^{-4}$. [[CO3](Apply/IOCQ)]
- (b) What is ozone depleting substance? Deduce the chemical formula of CFC-115. Write a short note on 'Montreal protocol'. [[CO3](Remember/LOCQ)]
- (c) What do you understand by suspended particulate matter? Describe its sources of formation and harmful effect on environment. [[CO3](Remember/LOCQ)]
3 + (1 + 2 + 2) + (2 + 2) = 12

Group - D

6. (a) Explain how a newly formed lake gradually becomes a Eutrophic lake. [[CO1](Remember/LOCQ)]
- (b) Why seeded water is used for dilution in a BOD test? Why the BOD₅ test is run in dark? [[CO2](Remember/LOCQ)]
- (c) Explain briefly the steps involved in surface water treatment to make it potable. [[CO4](Remember/LOCQ)]
- (d) What is the main purpose of using tertiary treatment of water? Write suitable reaction for separation of phosphate from water? [[CO3](Remember/LOCQ)]
3 + (1 + 2) + 3 + (2 + 1) = 12
7. (a) Why the heavy metals (As, Pb, Hg, Cd) are called nephrotoxin? Write the harmful effects of one of these heavy metals contaminated in water. [[CO1](Remember/LOCQ)]
- (b) What is the necessity of a five-day BOD test? The BOD₅ of a waste water sample is 200 mg/L and its ultimate BOD is 400 mg/L. Find the reaction rate constant. [[CO2](Apply/IOCQ)]
- (c) Write a short note on 'Waste Water Treatment'. [[CO4](Remember/LOCQ)]
(2 + 2) + (2 + 2) + 4 = 12

Group - E

8. (a) What is meant by hazardous wastes? Discuss briefly about the chemical treatment of hazardous wastes. [[CO5](Remember/LOCQ)]
- (b) What do you mean by noise pollution? What are the effects of noise pollution on living organism? [[CO6](Remember/LOCQ)]
- (c) Calculate the intensity of 101dB sound. (Reference intensity = $1 \times 10^{-12} \text{ W/m}^2$). [[CO2](Apply/IOCQ)]
- (d) How does modern agricultural practice affect the soil pollution? [[CO6](Remember/LOCQ)]
(1 + 3) + (1 + 3) + 2 + 2 = 12
9. (a) If the intensity of sound source is increased to four times the earlier intensity, then find out the increases in intensity level. [[CO3](Apply/IOCQ)]

- (b) What is land pollution? Write down the effects of industrial solid waste on land pollution. *[(CO4)(Remember/LOCQ)]*
- (c) What is noise threshold limit value? What are the drawbacks of dBA scale? *[(CO5)(Remember/LOCQ)]*
- (d) What is vermicomposting? *[(CO6)(Remember/LOCQ)]*
- 3 + (1 + 3) + (2 + 1) + 2 = 12**
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Cognition Level	LOCQ	IOCQ	HOCQ
Percentage distribution	75	25	0